Name:	Date:	Period:

QUESTIONS FOR CHEMMATTERS ARTICLE: "LINUS PAULING: AMERICAN HERO"

Read the article, then answer these questions on a piece of notebook paper.

- 1. At the time Linus Pauling was learning chemistry in college, what was the current and widely accepted model of chemical bonding called and how did it work?
- 2. Pauling thought that rather than the black and white definition of ionic and covalent bonds, bonds could be described on a continuum. What is the reason Pauling used to explain this theory?
- 3. What is the weak interaction among the N-H group of one amino acid and the C=O group of another amino acid called? What affect does this weak interaction have on the shape of the protein chain?
- 4. What, specifically, was Pauling's main concern about atomic bombs: e.g., the sheer magnitude of the destruction, the large amounts of heat given off in the explosion, something else entirely?
- 5. What were the consequences of his anti-nuclear, anti-war stance?

6. To what areas of human endeavor did Pauling contribute significantly?

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